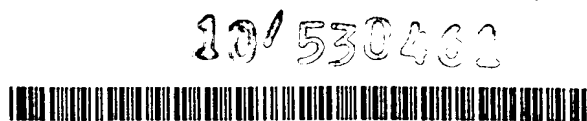


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(54) Title: HIGH-VOLTAGE DIRECT CURRENT CABLE INSULATION AND SEMICONDUCTIVE SHIELD

(57) Abstract: A high-voltage direct current cable insulation is made from a blend which includes an ethylene copolymer, such as ethylene-alpha olefin copolymer, with low crystallinity to reduce physical space charge trapping sites, a polar polymer modifier in an effective amount to enhance local conductivity to leak space charge quickly when local stress is enhanced, and an ion scavenger to stabilize or neutralize the space charge to provide a composition which is an effective high-voltage DC cable insulation. A high-voltage direct current cable semiconductive shield is made from a blend that includes an ethylene copolymer, a carbon black having a low level of ionic species, a polar polymer modifier, and an ion scavenger.

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HIGH-VOLTAGE DIRECT CURRENT
CABLE INSULATION AND SEMICONDUCTIVE SHIELD

FIELD OF THE INVENTION

This invention is directed to insulation and a semiconductive shield for power cables. More particularly, this invention is directed to insulation and a semiconductive shield for high-voltage direct current power cables.

DESCRIPTION OF THE PRIOR ART

Direct Current (DC) power transmission has several advantages over alternating current (AC) power transmission. DC transmission does not have a length limit, permits long-distance submarine cables (>50 km), has good connectivity among different networks/sources (such as windmills), has lower operating costs due to low conductor loss and no power loss, has superior power quality and flow control for system reliability/stability, and has higher voltage ratings. Cables insulated with oil/paper insulation have been successfully used for high-voltage direct current (HVDC) applications since 1954. Cables insulated with crosslinked polyethylene can have several advantages over cables insulated with oil/paper for HVDC applications. The advantages of crosslinked polyethylene include lower manufacturing costs, lower operation costs, easier maintenance for utilities, higher temperature ratings (such as 90 degrees C vs. 60 degrees C to 70 degrees C) to utilities, and environmental friendliness due to no oil leakage.

Polymeric dielectric insulating materials, particularly polyethylene without modification, however, cannot be used for HVDC applications. These materials have local space charge buildup, which can significantly enhance local fields under surge or lightning impulse, have charge neutralizations during reverse polarity, which can reduce local DC breakdown strength, and have stress inversions due to temperature-dependent conductivity, which can reverse local field enhancement.

A known approach to develop HVDC polymeric cable insulation products has been to have low and well-distributed space charge traps. Space charge can be trapped by physical traps formed between crystallinity and amorphous boundaries or chemical traps due to chemical structures of substances. The instant invention, however, is a cable insulation made from a blend which includes an ethylene copolymer, such as an ethylene-alpha olefin copolymer with low crystallinity to reduce physical space charge

trapping sites. The invention uses at least one polar polymer modifier in an effective amount to enhance local conductivity to leak space charge quickly when local stress is enhanced, and at least one ion scavenger to stabilize or neutralize the space charge to provide a composition which is an effective high-voltage DC cable insulation. The instant invention is also a semiconductive shield made from a blend that includes an ethylene copolymer, a carbon black having low levels of ionic species, a polar polymer modifier, and an ion scavenger.

SUMMARY OF THE INVENTION

The invention is directed to (1) a direct current cable, which includes insulation, which resists breakdown and deterioration when exposed to high-voltage direct current, (2) an insulation composition which resists deterioration and breakdown when exposed to high-voltage direct current, and (3) a method for reducing the deterioration of such insulation. The invention is also directed to a semiconductive shield with similar resistance to breakdown and deterioration.

The cable insulation composition includes at least one crosslinked nonpolar, low crystallinity resin with a density of less than 0.900grams/cubic centimeter which tends not to trap charge or create charge trap sites for a cable insulation temperature rating of at least 90degrees C. In another aspect, the resin is not crosslinked or is crosslinked only in a low amount (hereinafter a non-crosslinked polymer) which is effective for providing a cable insulation with a temperature rating of 75degrees C or above. In either aspect, the cable insulation also includes (1) at least one polar polymeric modifier which dissipates or leaks charge quickly under high fields, (2) at least one ion scavenger which stabilizes or neutralizes space charges, and (3) optionally at least one heat stabilizer which minimizes internal charge generation during in service thermal degradation of insulation.

The crosslinked nonpolar low crystalline resin, polar polymeric modifier, ion scavenger and heat stabilizer are in amounts effective for achieving temperature rating of 90degrees C or above, a charge density less of than 2 Coulomb/mm³ measured by a pulsed electro acoustic (PEA) method after 24 hours with either positive or negative 20 kV/mm applied. For the cable insulation which has a temperature rating of not more than 75degrees C the amount and extent of crosslinking of such resin, the amounts of polar polymeric modifier, ion scavenger and heat stabilizer all are effective for

achieving temperature rating of 75degreesC or above, a charge density less of than 2 Coulomb/mm³ measured by a pulsed electro acoustic (PEA) method after 24 hours with either positive or negative 20 kV/mm applied.

In another aspect, the invention is a high-voltage direct current cable insulation composition which has a temperature rating of 90degreesC or above and which comprises a blend of or which is made from a blend of at least one crosslinked ethylene copolymer, such as ethylene/alpha olefin copolymer, having a density of less than 0.900grams/cubic centimeter, a melt index of from 0.5 to 10grams/10 minutes, a crystallinity of less than about 10 percent; at least one polar polymeric modifier in an amount effective to provide field conductivity and permitting leakage of space and charge only at high fields; at least one ion scavenger in an amount effective to reduce charge build-up relative to a blend which does not include an ion scavenger; and, optionally, at least one heat stabilizer in an amount effective to prevent thermally induced degradation and resulting internal charge generation. The polar polymeric modifier, ion scavenger, and optional heat stabilizer are in amounts and ratios which when in combination with the crosslinked resin provide the insulation with a charge density less than 2 Coulomb/mm³ measured by a PEA method after 24 hours with either positive or negative 20 kV/mm applied.

In another aspect, for cable insulation which has a temperature rating of 75degreesC or above, the cable insulation composition comprises a nonpolar, non-crosslinked ethylene copolymer, such as an ethylene/alpha olefin copolymer, having a density of less than 0.900 grams/cubic centimeter a melt index of from 0.5 to 10grams/10 minutes, a crystallinity of less than about 10 percent; at least one polar polymeric modifier in an amount effective to provide field conductivity and permitting leakage of space and charge only at high fields; at least one ion scavenger in an amount effective to reduce charge build-up relative to a blend which does not include an ion scavenger; and, optionally, at least one heat stabilizer in an amount effective to prevent thermally induced degradation and resulting internal charge generation. The polar polymeric modifier, ion scavenger, and optional heat stabilizer are in amounts and ratios which when in combination with the resin provide the insulation with a charge density less than 2 Coulomb/mm³ measured by a PEA method after 24 hours with either positive or negative 20 kV/mm applied.

In yet another aspect, the invention is a high-voltage direct current cable insulation which comprises a blend of or which is made from a blend of at least one crosslinked ethylene/butene or ethylene/hexene olefin polymer having a density of less than 0.900grams/cubic centimeter, a melt index of from 0.5 to 10grams/10 minutes; from 0.1 to 15 weight percent of at least one polar polymeric modifier; from 0.05 to 0.5 weight percent of at least one charge scavenger to reduce charge build-up, and optionally, from 0.1 to 5 weight percent of at least one heat stabilizer in an amount effective to prevent thermally induced degradation and resulting internal charge generation.

The semiconductive shield composition of the present invention includes (a) at least one nonpolar, low crystallinity resin with a density of less than 0.900grams/cubic centimeter, (b) a carbon black having low levels of ionic species, (c) at least one polar polymeric modifier, and (d) at least one ion scavenger. Optionally, the composition can include at least one heat stabilizer. The resin can be crosslinked or not. The polar polymeric modifier dissipates or leaks charge quickly under high fields. The ion scavenger stabilizes or neutralizes space charges. The optional heat stabilizer minimizes internal charge generation during in service thermal degradation of insulation. The resulting cable should achieve either a temperature rating of (a) 90degrees C or above or (b) 75degrees C or above.

BRIEF DESCRIPTION OF DRAWING

Figure 1 describes PEA space charge measurements after 24 hours at +20 kV/mm.

Figure 2 describes PEA space charge measurements after 24 hours at -20 kV/mm.

DESCRIPTION OF THE INVENTION

The nonpolar ethylene copolymer, which can be used in the invention, includes ethylene/alpha olefin interpolymers, such as an ethylene/propylene copolymer. The resin has low crystallinity and has a density of less than 0.90grams/cubic centimeter. In a very important aspect, the resin used in the invention is a C₂-C₆ alpha olefin copolymer. Low crystallinity means a crystallinity of less than 20 percent as determined by a differential scanning calorimeter. The alpha olefin resins, which may be used in the invention, include an ethylene-hexene copolymer made with a single site

catalyst (SSC), an ethylene-butene copolymer made with a Ziegler Natta (Z/N) catalyst, and an ethylene-octene copolymer made with a SSC catalyst. The nonpolar ethylene copolymer may have some polar components, but such polar components should not be in such an amount to make the resin crystalline and lose its amorphous characteristics.

5 Hence, the nonpolar resin may contain an ethylene/styrene copolymer, an ethylene vinyl acetate copolymer, or an ethylene/ethyl acrylate copolymer in low amounts. In the aspect of the invention, which includes a crosslinked resin, the resin may be crosslinked using a peroxide, irradiation or a moisture cure.

Polar polymer modifiers are polymeric materials having at least one polar
10 component. These polar components may be a part of the polymer structure as side groups which group may be residues of maleic anhydride, vinyl acetate and vinyl acrylate, where such compounds have been incorporated into the polymer, such as by grafting or were a part of the monomer precursor of the polymer. Polar components also may include hydroxyl group, styrenic group and carboxyl group. The polar
15 polymeric modifier may be polyethylene glycol (where the polar component is hydroxyl group), ethylene ethyl acrylate (where the polar component is a residue of vinyl acrylate), ethylene styrene copolymer (where the polar component is a styrenic group) or a polyester having an acid number (where the polar component is a carboxyl group). The polar polymer modifiers may include maleic-anhydride-grafted very low
20 density ethylene/alpha olefin copolymers having a density of less than about 0.900grams/cubic centimeter as described above having about 0.3 percent maleic anhydride, polycaprolactone resins (having a carboxyl group in the main chain with a diol group at the end) and mixtures thereof.

Ion scavengers are compounds that have chelating groups, such as hydroxyl and
25 carboxyl. Ion scavengers may include 1,2-bis(3,5-di-tert-butyl-4-hydroxyhydrocinnamoyl) hydrazine, poly[[6-[1,1,3,3-tetramethylbutyl]amino]-s-triazine-2,4-diyl] [2,2,6,6-tetramethyl-4-piperidyl]imino]hexamethylene [(2,2,6,6-tetramethyl-4-piperidyl)imino], N,N'-bis(0-hydroxybenzal) oxalyldihydride, barbituric acid, tertiary phosphorous acid ester of a thiobisphenol, and N,N'-diphenyloxamid,
30 and mixtures thereof.

Antioxidants also may be put into the insulation or semiconductive shield compositions. Antioxidants, which may be used, include: 1,3,5-tris(4-tert-butyl-3-

hydroxy-2,6-dimethylbenzyl)-1,3,5-triazine-2,4,6-(1H, 3H, 5H)-trione, commercially available as Cyanox 1790; and distearylthiodipropionate (DSTDTP).

For semiconductive shield compositions, the carbon black should have a low level of ionic species, preferably less than about 200 ppm. More preferably, the amount of ionic species is less than about 100 ppm. The amount of ionic species of a carbon black can be determined by Induction Coupling Plasma Spectroscopy or the method described in J. Tanaka, "Interfacial Aging Phenomena In Power Cable Insulation systems", Institute of Materials Science, University of Connecticut, Progress Report No. 8 and 9, September 13, 1988.

For a crosslinked insulation composition with a temperature rating of 90degrees C, its elongation and set at a temperature of 150degreesC per ICEA T-28-562 test method should not be greater than 175 percent and 10 percent, respectively. The alternative referee method is the solvent extraction test per ASTM D2765. The crosslinked insulation composition generally will have maximum extractables after 20 hours drying time of no more than 30 percent. Insulation with a temperature rating of 75degreesC generally requires having percent retained tensile strength and elongation at break of no less than 70 percent after heat aged at 113degreesC for 7 days in air-circulated over per UL-1581 standard.

Examples 1-7

Examples 1, 2, 3, 4 and 6 illustrate the invention. Examples 5 and 7 are comparative examples.

For each example, the base ethylene polymer was characterized as having low crystallinity and a low melt index and as being a very low density polyethylene: (a) Exact 4033™ ethylene/hexene copolymer; (b) DGH-8480™ ethylene/butene copolymer; or (c) Engage 8003™ ethylene/octene copolymer. Unless otherwise indicated in Table 1, the exemplified composition contained Exact 4033™ ethylene/hexene copolymer as the base polymer.

Exact 4033™ ethylene/hexene copolymer, having a density of 0.880 grams/cubic centimeter and a melt index of 0.8 grams/10 minutes, is a single-site catalyzed polyethylene available from Exxon Chemical Co. DGH-8480™ ethylene/butene copolymer, having a density of 0.884 grams/cubic centimeter and a melt index of 0.8 grams/10 minutes, is available from The Dow Chemical Company.

Engage 8003™ ethylene/octene copolymer, having a density of 0.885 grams/cubic centimeter and a melt index of 1.0 grams/10 minutes, is a single-site catalyzed polyethylene available from DuPont Dow Elastomers LLC.

5 All of the exemplified compositions also contained 0.25 weight percent of Chimassorb 944 poly[[6-[1,1,3,3-tetramethyl-butyl)amino]-s-triazine-2,4-diyl] [2,2,6,6-tetramethyl-4-piperidyl)imino]hexamethylene[(2,2,6,6-tetramethyl-4-piperidyl)imino]] as an ion scavenger, 0.14 weight percent of Cyanox 1790 1,3,5-tris(4-tert-butyl-3-hydroxy-2,6-dimethylbenzyl)-1,3,5-triazine-2,4,6-(1H, 3H, 5H)-trione as a primary antioxidant, and 0.23 weight percent of DSTDP as a secondary antioxidant.
10 Chimassorb 994 is available from Ciba Specialty Chemicals Corporation. Cyanox 1790 is available from Cytec Corporation. DSTDP is available from Great Lakes Corporation.

Also, each composition was cured with bis(1-methyl-1-phenylethyl) peroxide, which is available from Hercules Corporation.

15 Various other components were used in the exemplified composition. DEFA-1373™ very low density ethylene/butene copolymer, having a 0.3 weight percent maleic anhydride graft, is available from The Dow Chemical Company and characterized as a polar polymer modifier. DEFA-1373 has a density of 0.903 grams/cubic centimeter and a melt index of 2.0 grams/10 minutes. Tone Polymer P-
20 767™ polylactone resin has a density of 1.145 grams/cubic centimeter, a melt index of 30.0 grams/10 minutes, and melting point of 60 degrees C. P-767 polylactone resin is available from The Dow Chemical Company and characterized as a polar polymer modifier. Zinc oxide, which was added as a heat stabilizer/phonon dissipator, is available as Kadox 911P from Zinc Corporation of America. Irganox 1024 1,2-bis(3,5-
25 di-tert-butyl-4-hydroxyhydrocinnamoyl)hydrazine, which was added as an ion scavenger, is available from Ciba Specialty Chemicals Corporation.

The space charge measurements were performed by a pulsed electro acoustic method. The details of this method can be found in literature as described in Y. Li, M. Yasuda, and T. Takad, "*Pulsed Electro-acoustic Method for Measurement of Charge*
30 *Accumulation in Solid Dielectrics*," IEEE Transaction EI, Vol. 1, pp. 188-195, 1994.

Each sample had 1.6mm thickness with a diameter of 135mm, placed between semicon electrodes of 0.1mm and a diameter of 30mm, placed between semicon

electrodes of 0.1mm and diameter of 30mm. The application of 32 kV DC (20 kV/mm) was applied for 24 hours, and space charge was measured by PEA without voltage applied as shown in Figure 1. The sample was grounded without applied voltage for 12 hours, and then voltage was applied with -32 kV DC (20 kV/mm) for 24 hours. The space charge without voltage applied was measured again by the PEA as shown in Figure 2. All measurements were done at ambient temperature about 20degrees C. Space charge measurements were plotted as charge density (Coulomb per cubic millimeter) as a function of time (nano-second). Each division shown in Figures 1 and 2 is equivalent to a value of 2 Coulomb/mm³.

For HVDC cable applications, HVDC cable insulation should keep the space charge as low as possible and as uniform as possible throughout the measurement of time. The value of space charge measurement for excellent HVDC cable insulation should be no more than 2 Coulomb/mm³ for both positive and negative DC stress.

TABLE I. HVDC INSULATION

Component	Ex. 1	Ex. 2	Ex. 3	Ex. 4	C. Ex. 5	Ex. 6	C. Ex. 7
base ethylene polymer	89.28	97.28	95.28	85.28	99.38	97.28 ¹	97.28 ²
maleic anhydride-grafted polyethylene	10.00			10.00			
polylactone resin		2.00	2.00	2.00		2.00	2.00
zinc oxide			2.00	2.00			
Irganox 1024	0.10	0.10	0.10	0.10		0.10	0.10
Weight percent ratio of thermoplastics to curing agent	99.2 : 0.8	99.2 : 0.8	99.2 : 0.8	99.2 : 0.8	99.2 : 0.8	99.0 : 1.0	99.0 : 1.0

¹ DGH-8480™ ethylene/butene copolymer.² Engage 8003™ ethylene/octene copolymer.

Effect of Additives

Comparative Example 5 containing typical antioxidants and UV stabilizer did not meet the desired requirement on space charge value at the applied positive DC stress of 20 kV/mm. However, Examples 1 and 2 with Irganox 1024 and two different polar polymer modifiers, respectively, met the desired requirements at both positive and negative DC stresses. Example 2 showed lower space charge distribution than Example 1. Example 3 with additional heat stabilizer, zinc oxide, showed further improvement in space charge when compared with Example 2. Example 4 with the combination of additive packages from Example 3 and 1 showed acceptable space charge performance.

Effect of the Resins

Examples 2 and 6 and Comparative Example 7 showed the effect of various VLDPE resins on space charge distribution. Comparative Example 7 made by octene comonomer did not meet the space charge distribution criteria with the levels of polymer modifier and ion scavenger shown.

What is claimed is:

- 1 1. High-voltage direct current cable insulation comprising:
 - 2 a blend of or which is made from a blend of
 - 3 (a) at least one ethylene copolymer, having a density of less than about
4 0.900grams/cubic centimeter, a melt index of from about 0.5 to about
5 10grams/10 minutes, a crystallinity of less than about 10 percent and a
6 catalyst residue of less than about 1000 ppm, selected from the group
7 consisting of
 - 8 (i) ethylene/alpha olefin copolymers and
 - 9 (ii) nonpolar, low crystalline ethylene copolymers selected from the
10 group consisting of ethylene/propylene copolymer and
11 ethylene/styrene copolymer and mixtures thereof;
 - 12 (b) at least one polar polymer modifier in an amount effective to provide an
13 insulation made with the blend with an enhanced field conductivity and
14 enhanced space charge leakage at high fields relative to an insulation
15 made with a blend which does not include a polar polymer modifier; and
 - 16 (c) at least one ion scavenger in an amount effective to reduce ionic
17 mobility relative an insulation made with a blend which does not include
18 an ion scavenger,
 - 19 wherein the ethylene copolymer, the polar polymer modifier, and the ion scavenger
20 being in amounts to provide the cable insulation with a charge density of less than 2
21 Coulomb/mm³ measured by a pulsed electro acoustic method after 24 hours with either
22 positive or negative 20 kV/mm.
- 1 2. High-voltage direct current cable insulation comprising:
 - 2 a blend of or which is made from a blend of
 - 3 (a) at least one ethylene/alpha olefin copolymer having a density of less
4 than about 0.900grams/cubic centimeter, a melt index of from about 0.5
5 to about 10grams/10 minutes, a crystallinity of less than about 10
6 percent and a catalyst residue of less than about 1000 ppm;
 - 7 (b) from about 0.1 to about 15 weight percent of at least one polar polymer
8 modifier having at least one polar component; and

9 (c) from about 0.05 to about 0.5 weight percent of at least one ion scavenger
10 having at least one chelating component,
11 wherein the ethylene/alpha olefin copolymer, the polar polymer modifier, and the ion
12 scavenger being in amounts to provide the cable insulation with a charge density of less
13 than 2 Coulomb/mm³ measured by a pulsed electro acoustic method after 24 hours with
14 either positive or negative 20 kV/mm applied.

1 3. The high-voltage direct current insulation of claim 1 or 2, wherein

2 (a) the polar polymer modifier is selected from the group consisting of (i) a
3 polymer having a density of less than 0.900grams/cubic centimeter with
4 at least one side group selected from the group consisting of hydroxyl,
5 carboxyl, styrenic; (ii) a polymer having a density of less than
6 0.900grams/cubic centimeter and at least one side group which is a
7 residue of maleic anhydride, vinyl acetate or vinyl acrylate; (iii) a
8 polylactone resin and; (iv) mixtures thereof, and

9 (b) the ion scavenger has at least one chelating group.

1 4. The high-voltage direct current insulation of any of claims 1 - 3, wherein the
2 ethylene copolymer is crosslinked.

1 5. A high-voltage direct current cable comprising:

2 (a) an electrical conductor; and

3 (b) cable insulation comprising

4 a blend or which is made from a blend of

5 (i) at least one nonpolar, low crystalline ethylene copolymer
6 selected from the group consisting of ethylene/propylene
7 copolymer and ethylene/styrene copolymer and mixtures thereof,
8 the ethylene copolymer having a density of less than about
9 0.900grams/cubic centimeter, a melt index of from about 0.5 to
10 about 10grams/10 minutes, a crystallinity of less than about 10
11 percent and a catalyst residue of less than about 1000 ppm;

12 (ii) at least one polar polymer modifier having at least one polar
13 component in an amount effective to provide an insulation made
14 with the blend with an enhanced field conductivity and enhanced
15 space charge leakage at high fields relative to an insulation made

16 with a blend which does not include a polar polymer modifier;
17 and
18 (iii) at least one ion scavenger having at least one chelating
19 component in an amount effective to reduce ion mobility relative
20 to an insulation made with a blend which does not include an ion
21 scavenger,

22 wherein the ethylene copolymer, the polar polymer modifier, and the ion scavenger
23 being in amounts to provide the cable insulation with a charge density of less than 2
24 Coulomb/mm³ measured by a pulsed electro acoustic method after 24 hours with either
25 positive or negative 20 kV/mm applied.

1 6. The high-voltage direct current cable as recited in claim 5 wherein the ethylene
2 copolymer is crosslinked.

1 7. A method for providing a cable insulation with a charge density of less than 2
2 Coulomb/mm³ measured by a pulsed electro acoustic method after 24 hours with either
3 positive or negative 20 kV/mm applied, the method comprising:

4 (a) mixing

5 (i) at least one ethylene/alpha olefin copolymer having a density of
6 less than about 0.900grams/cubic centimeter, a melt index of
7 from about 0.5 to about 10grams/10 minutes, a crystallinity of
8 less than about 10 percent and a catalyst residue of less than
9 about 1000 ppm.

10 (ii) from about 0.1 to about 15 weight percent of at least one polar
11 polymer modifier having at least one polar component;

12 (iii) from about 0.05 to about 0.5 weight percent of at least one ion
13 scavenger having at least one chelating component,

14 wherein the ethylene/alpha olefin copolymer, the polar polymer modifier and the ion
15 scavenger being in amounts to provide the cable insulation with a charge density of less
16 than 2 Coulomb/mm³ measured by a pulsed electro acoustic method after 24 hours with
17 either positive or negative 20 kV/mm applied.

1 8. High-voltage direct current cable semiconductive shield comprising:
2 a blend of or which is made from a blend of

- 3 (a) at least one ethylene copolymer, having a density of less than about
4 0.900grams/cubic centimeter, a melt index of from about 0.5 to about
5 10grams/10 minutes, a crystallinity of less than about 10 percent and a
6 catalyst residue of less than about 1000 ppm, selected from the group
7 consisting of
8 (i) ethylene/alpha olefin copolymers and
9 (ii) nonpolar, low crystalline ethylene copolymers selected from the
10 group consisting of ethylene/propylene copolymer and
11 ethylene/styrene copolymer and mixtures thereof;
12 (b) a carbon black having a low level of ionic species;
13 (c) at least one polar polymer modifier in an amount effective to provide a
14 semiconductive shield made with the blend with an enhanced field
15 conductivity enhanced space charge leakage at high fields relative to a
16 semiconductive shield made with a blend which does not include a polar
17 polymer modifier; and
18 (d) at least one ion scavenger in an amount effective to reduce ionic
19 mobility relative to a semiconductive shield made with a blend, which
20 does not include an ion scavenger.
- 1 9. The high-voltage direct current semiconductive shield of claim 8, wherein
2 (a) the polar polymer modifier is selected from the group consisting of (i) a
3 polymer having a density of less than 0.900grams/cubic centimeter with
4 at least one side group selected from the group consisting of hydroxyl,
5 carboxyl, styrenic; (ii) a polymer having a density of less than
6 0.900grams/cubic centimeter and at least one side group which is a
7 residue of maleic anhydride, vinyl acetate or vinyl acrylate; (iii) a
8 polylactone resin and; (iv) mixtures thereof, and
9 (b) the ion scavenger has at least one chelating group.
- 1 10. The high-voltage direct current semiconductive shield of claim 8 or 9, wherein
2 the ethylene copolymer is crosslinked.

Druckexemplar

1/1

FIG. 1

Space Charge Measurement after 24 hours at + 20 kV/mm

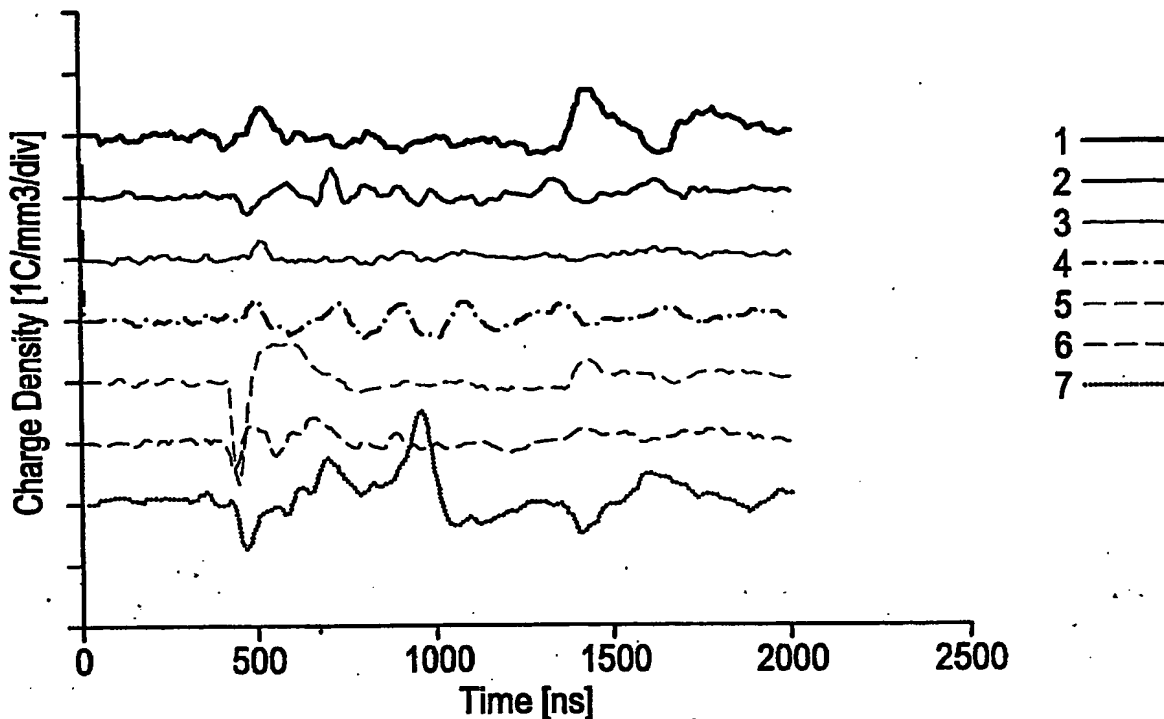
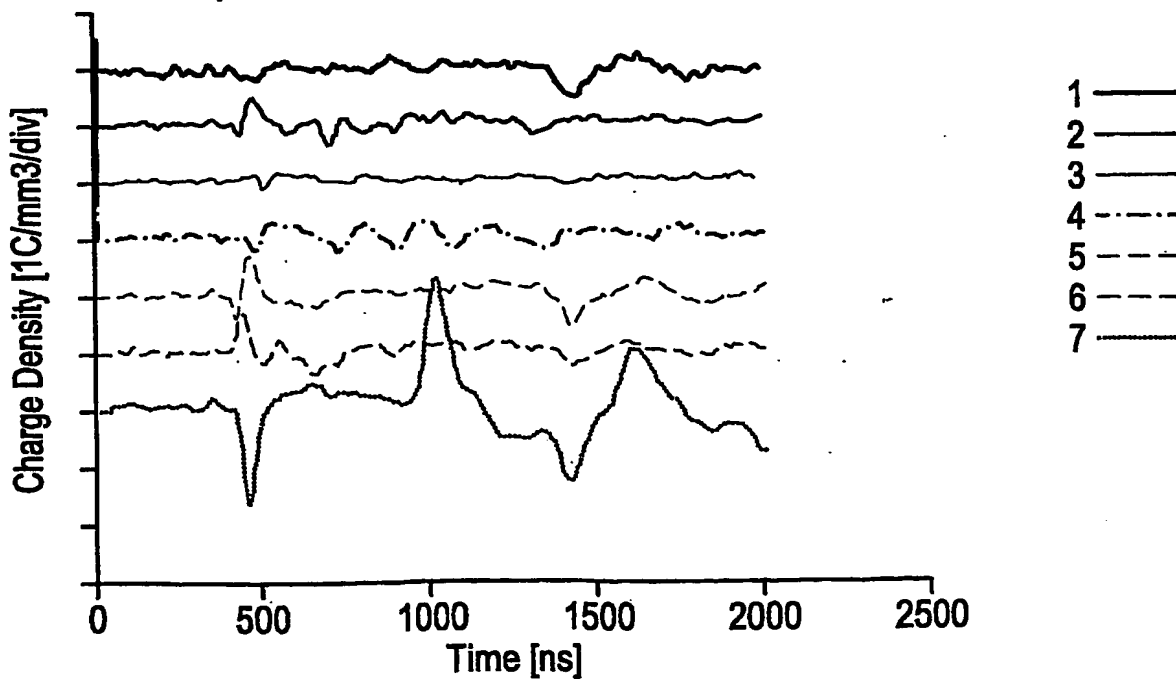


FIG. 2

Space Charge Measurement after 24 hours at - 20 kV/mm



1/1

FIG. 1

Space Charge Measurement after 24 hours at + 20 kV/mm

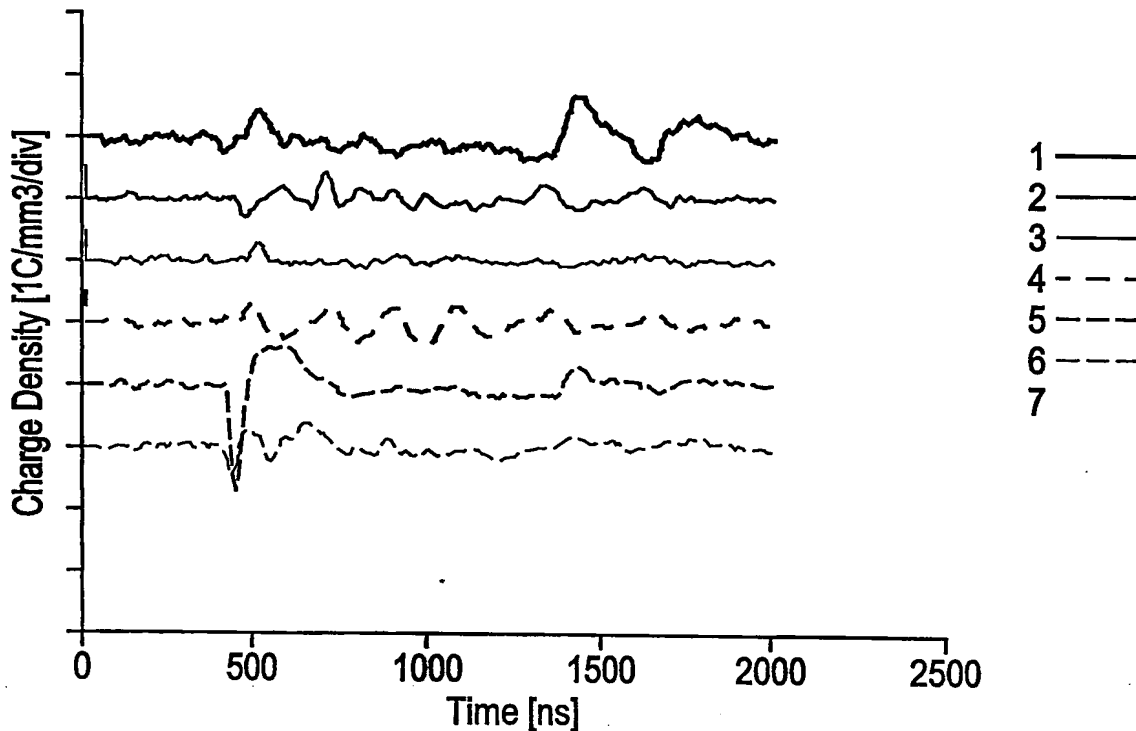
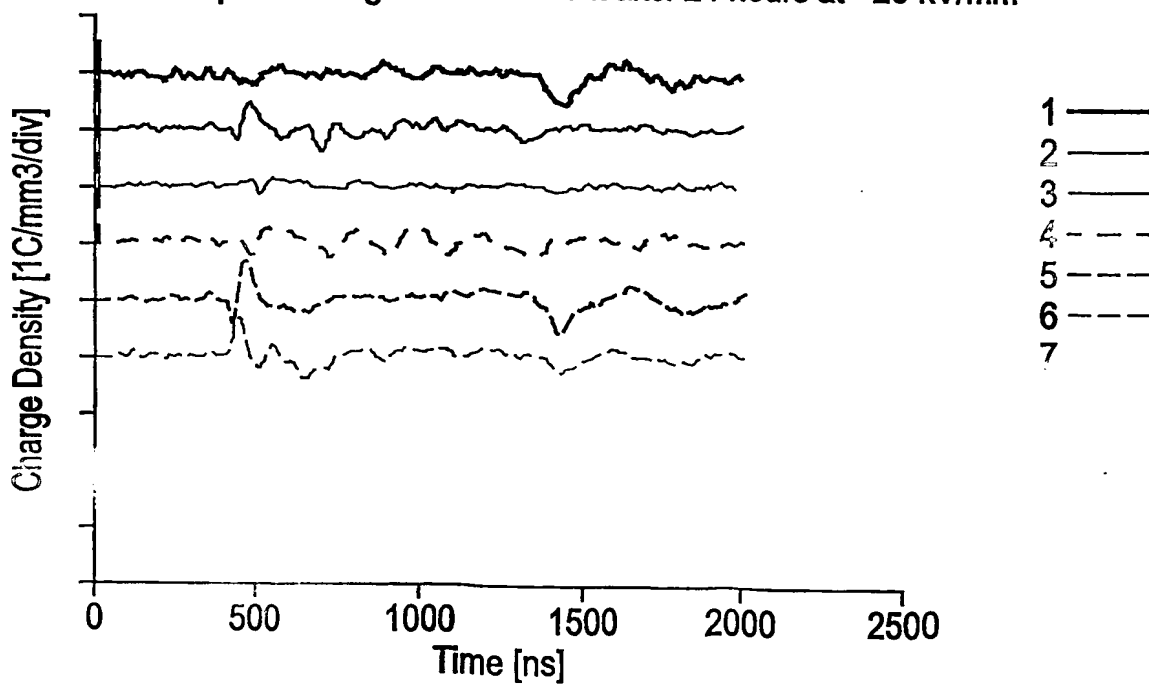


FIG. 2

Space Charge Measurement after 24 hours at - 20 kV/mm



INTERNATIONAL SEARCH REPORT

International Application No.

PCT/US 03/29070

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 H01B3/00 H01B3/18 H01B3/44 H01B1/24 C08L23/00

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 H01B C08L

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, CHEM ABS Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	WO 99 44206 A (ASEA BROWN BOVERI ;TOERNKVIST CHRISTER (SE); JONSSON JONAS (SE)) 2 September 1999 (1999-09-02) page 4, line 20 -page 5, line 4; claims 4,10,18	1-10
A	WO 99 40589 A (ASEA BROWN BOVERI ;CARSTENSEN PETER (SE)) 12 August 1999 (1999-08-12) page 4 -page 5	1-10
A	M. SALAH KHALIL: "International Research and Development Trends and Problems of HVDC Cables with Polymeric Insulation" IEEE ELECTRICAL INSULATION MAGAZINE, vol. 13, no. 6, 1997, page 35 XP001157058 page 39 -page 41	1-10

☐ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

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- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

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Date of the actual completion of the international search

19 January 2004

Date of mailing of the international search report

30/01/2004

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Marsitzky, D

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 03/29070

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